Regioselective Addition Reaction of Lithium Enolates to Thio-Substituted 1,4-Naphthoquinones. Convenient Synthesis of a Naphthofuran-4,9-dione Ring System

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The regioselective addition of lithium enolates to thiosubstituted 1,4-naphthoquinones gave alkylated 1,4-naphthoquinones via 1,4-addition products. A phenacyl 1,4-naphthoquinone was cyclized to a naphthofuran-4,9-dione ring system.

It is of great interest to achieve the selective 1,2- or 1,4-addition of carbon nucleophiles to quinones since introduction of a functionalized alkyl chain into the quinone skeleton is one of the most attractive synthetic route to biologically important naturally-occurring quinones. (1) Reactions of quinones with tin(II) enolates or silyl enol ethers have been reported to give 1,2- or 1,4-addition products, (2) whereas the Michael addition of lithium enolates to quinones has been less studied. (3) Here we wish to report the successful 1,4-addition of lithium enolates to thio-substituted 1,4-naphthoquinones (1).

Reaction of 2-phenylthio-1,4-naphthoquinone ($\underline{1a}$) and 2-ethylthio-1,4-naphthoquinone ($\underline{1b}$) with lithium enolates $\underline{2a-c}$ gave 2-(2-oxoalkyl)-3-phenylthio-1,4-naphthoquinones $\underline{4a-d}$ in excellent yields, which were derived from the initially formed 1,4-addition products $\underline{3}$ through the air-oxidation as shown in Scheme 1. No formation of other addition products was observed in these reactions. On the other hand, reaction of 2-methyl-1,4-naphthoquinone ($\underline{1c}$) with lithium enolate $\underline{2c}$ afforded 2,3-dihydro-2-methyl-3-(2-oxo-2-phenylethyl)-1,4-naphthoquinone ($\underline{5}$) and 2,3-dihydro-2-methyl-2-(2-oxo-2-phenylethyl)-1,4-naphthoquinone ($\underline{6}$) in 48 and 32% yields, respectively. It should be noted that the

thio-substituted group at the 2-position of the parent 1,4-naphthoquinone ($\underline{1}$) has a crucial role to determine the regiochemistry in these reactions. The addition of lithium enolate $\underline{2b}$ to 2-butylthio-3-chloro-1,4-naphthoquinone ($\underline{1d}$) also occurred regioselectively at the 3-position to give naphthoquinone $\underline{4e}$ in 70% yield via elimination of a chloride ion. These results are summarised in Table 1.

Scheme 1.

Table 1. Reaction of Naphthoquinone with Lithium Enolates

Entry	<u>1</u>			<u>2</u>	Products ^{a)}
	R	X		R^1 R^2	(Yield/%) ^{b)}
1	SPh	Н	(<u>1a</u>)	-(CH ₂) ₃ - (<u>2a</u>)	<u>4a</u> (76)
2	SPh	Н	(<u>1a</u>)	-(CH ₂) ₂ - (<u>2b</u>)	<u>4b</u> (75)
3	SPh	Н	(<u>1a</u>)	H Ph (<u>2c</u>)	<u>4c</u> (62)
4	SEt	Н	(<u>1b</u>)	$-(CH_2)_3 - (2a)$	<u>4d</u> (64)
5	Me	Н	(<u>1c</u>)	H Ph (<u>2c</u>)	5 (48) and 6 (32)
6	SBu	C1	(<u>1d</u>)	$-(CH_2)_2 - (\underline{2b})$	<u>4e</u> (70)

- a) All the products gave satisfactory IR, NMR, and mass spectra.
- b) Isolated yields.

A typical procedure of this reaction is as follows: butyllithium (2 mmol, hexane solution) was added dropwise over 5 min to a stirred solution of dicyclohexylamine (2 mmol) in THF (8 ml) at -45 $^{\circ}$ C under argon. After the addition the mixture was cooled to -78 $^{\circ}$ C. A solution of a ketone (2 mmol) in THF was added slowly and the mixture was stirred until it became a clear solution (10-15 min). Then HMPA (1 ml) was added. After stirring for 10 min at -78 $^{\circ}$ C, the resulting solution was added to a solution of naphthoquinone $\underline{1}$ (2 mmol) in THF (15 ml) and HMPA (1 ml) at -70 $^{\circ}$ C. The reaction mixture was stirred for 15 min and then 5% hydrochloric acid (10 ml) was added. Purification of the crude mixture with column chromatography (silica gel, toluene-hexane, 1:1) gave naphthoquinone $\underline{4}$.

Associated with this successful 1,4-addition reaction of thio-substituted 1,4-naphthoquinones, we attempted the transformation of a product to a naphthofuran-4,9-dione ring system. Thus, $2-(2-oxo-2-phenylethy1)-3-phenylthio-1,4-naphthoquinone <math>\underline{4c}$ was brominated in acetic acid at room temperature for 12 h to give the crude brominated naphthoquinone $\underline{7}$ which was treated with triethylamine (1 equiv.) in benzene at room temperature for 10 min to give 2-phenylnaphtho[2,3-b]furan-4,9-dione (8) 6 ,7) in 45% yield.

Scheme 2.

Regioselective introduction of 2-oxoalkyl chains into thio-substituted 1,4-naphthoquinones could be achieved by use of lithium enolates. Cyclization of the resulting 2-(2-oxoalkyl)-3-phenylthio-1,4-naphthoquinones ($\underline{4}$) provides an efficient method for the synthesis of naphthofuran-4,9-dione derivatives. A further study on the synthesis of naturally-occurring cytotoxic naphthofuran-4,9-diones⁸) is now in progress.

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References

Y. Naruta, H. Uno, and K. Maruyama, Tetrahedron Lett., <u>22</u>, 5221 (1981);
 A. Fischer and G. N. Henderson, ibid., <u>21</u>, 701 (1980); H. Uno, J. Org. Chem., <u>51</u>, 350 (1986); T. Kometani and E. Yoshii, J. Chem. Soc., Perkin Trans. 1, <u>1981</u>, 1191; M. F. Semmelhack and A. Zask, J. Am. Chem. Soc. <u>105</u>, 2034 (1983);
 A. Takuwa, O. Soga, T. Mishima, and K. Maruyama, J. Org. Chem., <u>52</u>, 1261 (1987).

- 2) T. Mukaiyama, R. S. J. Clark, and N. Iwasawa, Chem. Lett., 1987, 479; T. Mukaiyama, Y. Sagawa, And S. Kobayashi, ibid., 1987, 2169; K. Maruyama, S. Tai, and H. Imahori, Bull. Chem. Soc. Jpn., 59, 1777 (1986).
- 3) The reaction of a lithium enolate derived from a lactone with benzoquinone was reported, see: G. A. Kraus and B. Roth, Tetrahedron Lett., <u>36</u>, 3129 (1977).
- 4) M. Iwao and T. Kuraishi, Tetrahedron Lett., 26, 6213 (1985).
- 5) The reaction of 2-bromo-1,4-naphthoquinone and 2-methoxy-1,4-naphthoquinone with lithium enolate <u>2a</u> gave many products from which the addition products corresponding to 4 could not be isolated.
- 6) Compound 9 had mp 246-247 °C (Lit. 7) mp 246.5-247.5 °C); 1 HNMR (CDCl₃) δ =7.24 (s, 1H), 7.53 (m, 3H), 7.80 (m, 2H), 7.90 (m, 2H), 8.26 (m, 2H).
- 7) S. C. Hooker and A. Steyermark, J. Am. Chem. Soc., <u>58</u>, 1202 (1936).
- 8) M. M. Rao and D. G. I. Kinston, J. Nat. Prod., <u>45</u>, 600 (1982); X. A. Dominguez, Planta Med., <u>49</u>, 63 (1983); C. C. Lopes, R. S. C. Lopes, A. V. Pinto, and P. R. R. Costa, J. Heterocycl. Chem., <u>21</u>, 621 (1984); C. L. Zani, A. B. D. Oliveira, and V. Snieckus, Tetrahedron Lett., <u>28</u>, 6561 (1987); E. Ghera, R. Maurya, and Y. Ben-David, ibid., <u>27</u>, 3935 (1986).

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